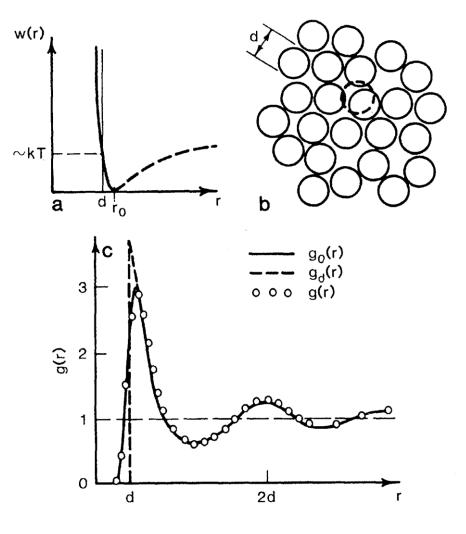
# Recapitulation: Density operators

- One particle  $n(r) = \langle \sum_i \delta(r r_i) \rangle$
- Two particle  $n(r,r') = \langle \sum_{i \neq j} \delta(r r_i) \delta(r' r_j) \rangle$ , the correlation of a particle with other particles.
- Pair correlation function  $g(r, r') = \frac{n(r, r')}{n(r)n(r')}$
- A typical W is the Lennard-Jones (L J)potential:

• 
$$w(r) = w_0 \left[ \left(\frac{r}{\sigma}\right)^{12} - \left(\frac{r}{\sigma}\right)^6 \right]$$

1



Recapitulation: the compressibility  $\chi$  and the equation of state:

$$n kT \chi = \frac{kT}{\partial p/\partial n} = 1 + n \int d^3r (g-1)$$
$$P = nkT - n^2 \int d^3r g(r)r \frac{\partial U(r)}{\partial r} / 6$$

Pair correlation function can be measured experimentally

- Pair correlation function  $g(r, r') = \frac{n(r, r')}{n(r)n(r')}$
- In scattering experiments, one can meassure

• 
$$S(k) = \sum \exp(ik \cdot r_{jl}) / N$$

• 
$$S(k) = \int \frac{drdr'}{N} < n(r)n(r') > e^{ik(r-r')} = 1 + n \left[\frac{\delta(k)}{N} + \int (g-1)e^{ik(r-r')}\right]$$

• The compressibility  $n kT \chi = \frac{kT}{\partial p/\partial n} = 1 + n \int d^3r(g-1)$  can be realted to the zero q limit of S(q)!

# Statistical physics of scattering, correlation, and structure



# **Scattering experiment**

#### Scattering experiment

Nature of experiment Interaction between probe and system Interaction matrix element Fourier transform of vOne-particle density Static structure factor Typical forms of  $S(\vec{q})$ Dilute gas Liquid structure factor Distribution function Critical point Crystal structure factor Reciprocal lattice Temperature variation Molecular dynamics Range of correlations Scattering divergence near critical point Compressibility rule Derivation of

compressibility rule One-particle density Average values Variance of N

Structure factor Significance

Pair distribution function

Let us now consider using scattering experiments to study a system described by the

$$H = \sum_{i=1}^{N} \frac{p_i^2}{2m} + \sum_{i < j} u(r_{ij}).$$



# **Scattering experiment**

#### Scattering experiment

Nature of experiment Interaction between probe and system Interaction matrix element Fourier transform of v

- One-particle density Static structure factor Typical forms of  $S(\vec{q})$ Dilute gas Liquid structure factor Distribution function Critical point Crystal structure factor Reciprocal lattice Temperature variation Molecular dynamics Range of correlations Scattering divergence near critical point
- Compressibility rule Derivation of compressibility rule One-particle density Average values
- Variance of N
- Pair distribution function
- Structure factor Significance

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■ We prepare a collimated beam of probe particles (neutrons, X-rays, etc.) with probe coordinates  $\vec{r_p}$  and momenta  $\hbar \vec{k}$ .



## **Scattering experiment**

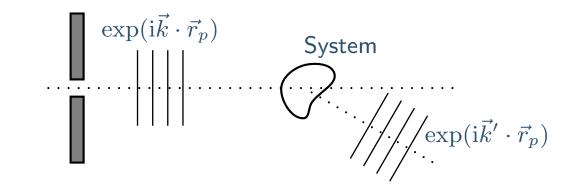
#### Scattering experiment

Nature of experiment Interaction between probe and system Interaction matrix element

- Fourier transform of vOne-particle density Static structure factor Typical forms of  $S(\vec{q})$ Dilute gas Liquid structure factor Distribution function Critical point Crystal structure factor **Reciprocal lattice** Temperature variation Molecular dynamics Range of correlations Scattering divergence near critical point Compressibility rule Derivation of compressibility rule One-particle density Average values Variance of NPair distribution function Structure factor Significance
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$$H = \sum_{i=1}^{N} \frac{p_i^2}{2m} + \sum_{i < j} u(r_{ij}).$$

- We prepare a collimated beam of probe particles (neutrons, X-rays, etc.) with probe coordinates  $\vec{r_p}$  and momenta  $\hbar \vec{k}$ .
- The incident beam is approximately in a plane-wave state  $\exp(i\vec{k}\cdot\vec{r}_p)$ :



The scattered beam is in a different plane-wave state  $\exp(i\vec{k}' \cdot \vec{r_p})$ , where  $\hbar \vec{k}'$  is the final momentum of the scattered probe particles.



### Static structure factor

Scattering experiment Nature of experiment Interaction between probe and system Interaction matrix element Fourier transform of vOne-particle density

#### Static structure factor

Typical forms of  $S(\vec{q})$ Dilute gas Liquid structure factor Distribution function Critical point Crystal structure factor **Reciprocal lattice** Temperature variation Molecular dynamics Range of correlations Scattering divergence near critical point Compressibility rule Derivation of compressibility rule One-particle density Average values Variance of NPair distribution function Structure factor Significance

### The experiments measure the static structure factor $S(q^{\rightarrow})$

$$S(\vec{q}) \equiv \frac{1}{N} \langle \hat{n}_1(\vec{q}) \hat{n}_1(-\vec{q}) \rangle - N \delta_{\vec{q},0},$$

in which the last term is introduced for later convenience (recall that the scattered beam cannot be measured at exactly  $q \rightarrow = 0$ ).

■  $S(q^{-})$  is related to the pair distribution function  $g(\vec{r})$  for a continuous phase by

$$S(\vec{q}) - 1 = \frac{N}{V} \int d^3r \left[ g(\vec{r}) - 1 \right] \exp(-i\vec{q} \cdot \vec{r}).$$

That is,  $S(\vec{q}) - 1$  is the Fourier transform of  $n[g(\vec{r}) - 1]$ .



Scattering experiment Nature of experiment Interaction between probe and system Interaction matrix element Fourier transform of vOne-particle density Static structure factor Typical forms of  $S(\vec{q})$ Dilute gas Liquid structure factor Distribution function Critical point

Crystal structure factor **Reciprocal lattice** Temperature variation Molecular dynamics Range of correlations Scattering divergence near critical point Compressibility rule Derivation of compressibility rule One-particle density Average values Variance of NPair distribution function Structure factor Significance

- What do we expect to obtain from measurements of  $S(\vec{q})$ ?
- Let us survey some typical results for gases, liquids, and crystals of spherically symmetric atoms.



Scattering experiment Nature of experiment Interaction between probe and system Interaction matrix element Fourier transform of v

One-particle density Static structure factor Typical forms of  $S(\vec{q})$ 

#### Dilute gas Liquid structure factor Distribution function Critical point Crystal structure factor **Reciprocal lattice** Temperature variation Molecular dynamics Range of correlations Scattering divergence near critical point Compressibility rule Derivation of compressibility rule One-particle density Average values Variance of NPair distribution function Structure factor Significance

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- Let us survey some typical results for gases, liquids, and crystals of spherically symmetric atoms.
- In dilute gases, we can use a virial expansion to write

$$g(r) = \exp[-\beta u(r)]\{1 + ng_1(r) + n^2g_2(r) + \cdots\}.$$



Scattering experiment Nature of experiment Interaction between probe and system Interaction matrix element

Fourier transform of vOne-particle density Static structure factor Typical forms of  $S(\vec{q})$ 

Dilute gas Liquid structure factor Distribution function Critical point Crystal structure factor **Reciprocal lattice** Temperature variation Molecular dynamics Range of correlations Scattering divergence near critical point Compressibility rule Derivation of compressibility rule One-particle density Average values Variance of NPair distribution function Structure factor Significance

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For hard spheres of diameter  $\sigma$  with no attractive interaction, the Fourier transform can be evaluated exactly to obtain<sup>2</sup>

$$S(q) = 1 - 4\pi n \left(\frac{\sin q\sigma - q\sigma \cos q\sigma}{q^3}\right) + O(n^2).$$



Scattering experiment Nature of experiment Interaction between probe and system Interaction matrix element

Fourier transform of vOne-particle density Static structure factor Typical forms of  $S(\vec{q})$ 

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$$S(q) = 1 - 4\pi n \left(\frac{\sin q\sigma - q\sigma \cos q\sigma}{q^3}\right) + O(n^2).$$

In the limit of an ideal gas (i.e.,  $n \rightarrow 0$ ), we have

 $S(q) \rightarrow 1.$ 

- Deviations of S(q) from 1 therefore give a measure of how far the system deviates from an ideal gas.
  - <sup>2</sup>N. W. Ashcroft and N. H. March, *Proc. Roy. Soc. London A* **297**, 336 (1967).



# **Dilute gas**

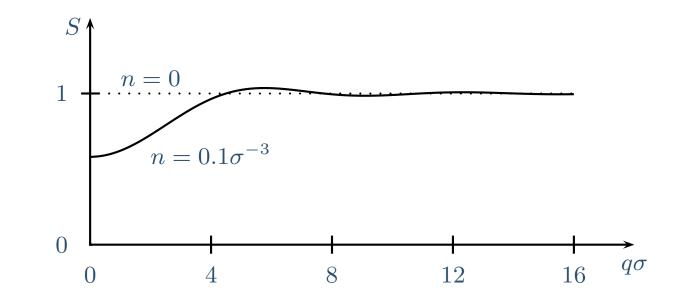
Scattering experiment Nature of experiment Interaction between probe and system Interaction matrix element

Fourier transform of vOne-particle density Static structure factor Typical forms of  $S(\vec{q})$ Dilute gas

Liquid structure factor Distribution function Critical point Crystal structure factor Reciprocal lattice Temperature variation Molecular dynamics Range of correlations Scattering divergence near critical point Compressibility rule Derivation of compressibility rule One-particle density Average values Variance of NPair distribution function Structure factor

Significance

The structure factor of a dilute gas of hard spheres is compared with that of an ideal gas in the figure below:



The structure factor of a dilute gas therefore drops below one for small values of q.



# Liquid structure factor

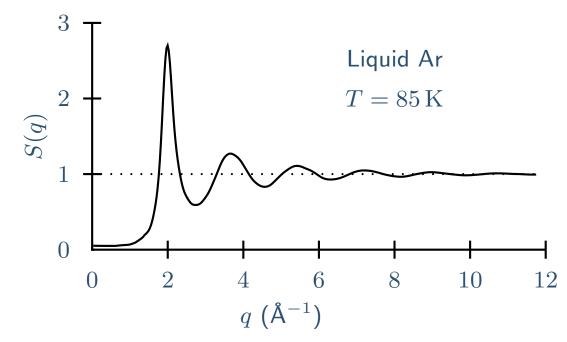
Scattering experiment Nature of experiment Interaction between probe and system Interaction matrix element Fourier transform of vOne-particle density

Static structure factor Typical forms of  $S(\vec{q})$  Dilute gas

Liquid structure factor Distribution function

Critical point Crystal structure factor **Reciprocal lattice** Temperature variation Molecular dynamics Range of correlations Scattering divergence near critical point Compressibility rule Derivation of compressibility rule One-particle density Average values Variance of NPair distribution function Structure factor Significance

By contrast, the measured structure factor of a liquid differs quite remarkably from that of an ideal gas:<sup>3</sup>



The data shown here were taken far away from the critical point  $(T_c = 151 \text{ K}, P_c = 48.1 \text{ atm}).$ 

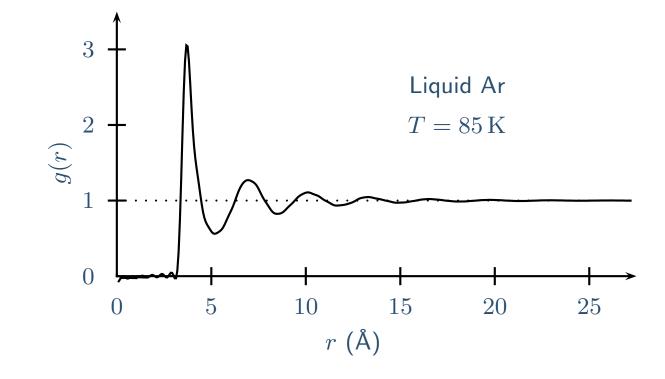
<sup>3</sup>J. L. Yarnell, M. J. Katz, R. G. Wenzel, and S. H. Koenig, *Phys. Rev. A* 7, 2130 (1973).



# **Distribution function**

Scattering experiment Nature of experiment Interaction between probe and system Interaction matrix element Fourier transform of vOne-particle density Static structure factor Typical forms of  $S(\vec{q})$ Dilute gas Liquid structure factor Distribution function Critical point Crystal structure factor **Reciprocal lattice** 

Temperature variation Molecular dynamics Range of correlations Scattering divergence near critical point Compressibility rule Derivation of compressibility rule One-particle density Average values Variance of *N* Pair distribution function Structure factor Significance The corresponding pair distribution function (obtained by Fourier transform) is shown below:



- The sharp peak at  $r \simeq 3.7$  Å shows that there is a strong tendency for atoms to cluster together at this separation in liquid argon.
- Subsidiary peaks at greater r show that the correlations extend beyond nearest neighbors.



# **Critical point**

Scattering experiment Nature of experiment Interaction between probe and system Interaction matrix element Fourier transform of v

One-particle density Static structure factor Typical forms of  $S(\vec{q})$ Dilute gas Liquid structure factor Distribution function

Critical point

Crystal structure factor Reciprocal lattice Temperature variation Molecular dynamics Range of correlations Scattering divergence near critical point Compressibility rule Derivation of compressibility rule One-particle density Average values Variance of *N* Pair distribution function Structure factor

Significance

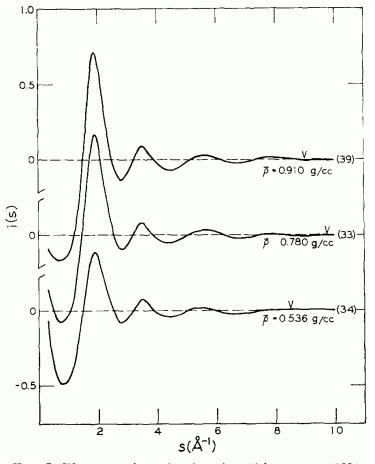


FIG. 5. The argon intensity function i(s) at  $t = -120^{\circ}$ C.

For a liquid close to the critical point, S(q) diverges near q = 0, becoming more divergent as  $(T, P) \rightarrow (T_c, P_c)$ .<sup>a</sup>



# **Critical point**

Scattering experiment Nature of experiment Interaction between probe and system Interaction matrix element Fourier transform of v

One-particle density Static structure factor Typical forms of  $S(\vec{q})$ Dilute gas Liquid structure factor Distribution function

Critical point

Crystal structure factor Reciprocal lattice Temperature variation Molecular dynamics Range of correlations Scattering divergence near critical point Compressibility rule Derivation of compressibility rule One-particle density Average values Variance of *N* Pair distribution function Structure factor

Significance

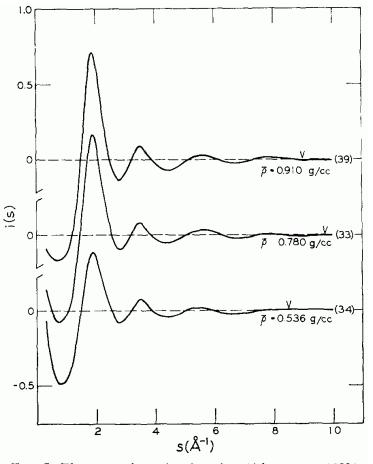


FIG. 5. The argon intensity function i(s) at t = -120 °C.

For a liquid close to the critical point, S(q) diverges near q = 0, becoming more divergent as  $(T, P) \rightarrow (T_c, P_c)$ .<sup>a</sup>

This says that the quantity

$$\frac{1}{N} \langle \hat{n}_1(\vec{q}) \hat{n}_1(-\vec{q}) \rangle$$

diverges as  $\vec{q} \to 0$  and  $(T, P) \to (T_c, P_c)$ .

Can you think of a reason why this would happen?

<sup>a</sup>P. G. Mikolaj and C. J. Pings, *J. Chem. Phys.* **46**, 1401 (1967).



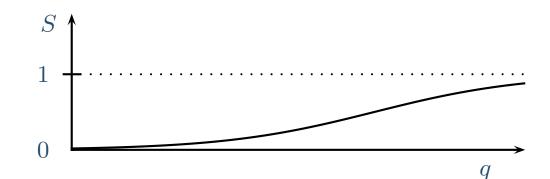
# **Crystal structure factor**

Scattering experiment Nature of experiment Interaction between probe and system Interaction matrix element

Fourier transform of vOne-particle density Static structure factor Typical forms of  $S(\vec{q})$ Dilute gas Liquid structure factor Distribution function Critical point

Crystal structure factor

Reciprocal lattice Temperature variation Molecular dynamics Range of correlations Scattering divergence near critical point Compressibility rule Derivation of compressibility rule One-particle density Average values Variance of *N* Pair distribution function Structure factor Significance For an arbitrary direction of  $\vec{q}$ , the structure factor of a crystal has the smooth form sketched below (due to thermal diffuse scattering by lattice vibrations):





# **Crystal structure factor**

Scattering experiment Nature of experiment Interaction between probe and system Interaction matrix element

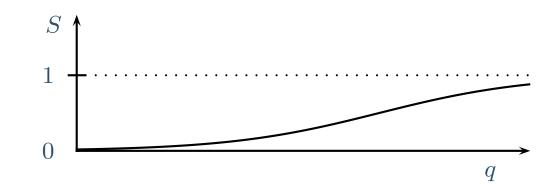
Fourier transform of vOne-particle density Static structure factor Typical forms of  $S(\vec{q})$ Dilute gas Liquid structure factor Distribution function Critical point

#### Crystal structure factor

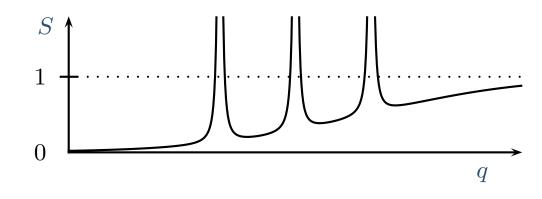
Reciprocal lattice Temperature variation Molecular dynamics Range of correlations Scattering divergence near critical point Compressibility rule Derivation of compressibility rule One-particle density Average values Variance of *N* Pair distribution function Structure factor

Significance

For an arbitrary direction of  $\vec{q}$ , the structure factor of a crystal has the smooth form sketched below (due to thermal diffuse scattering by phonons):



• However, for certain special directions of  $\vec{q}$ , there are sharp peaks (of order N):





# **Reciprocal lattice**

Scattering experiment Nature of experiment Interaction between probe and system Interaction matrix element Fourier transform of v

Fourier transform of vOne-particle density Static structure factor Typical forms of  $S(\vec{q})$ Dilute gas Liquid structure factor Distribution function

#### Critical point Crystal structure factor

#### Reciprocal lattice

Temperature variation Molecular dynamics Range of correlations Scattering divergence near critical point Compressibility rule Derivation of compressibility rule One-particle density Average values Variance of *N* Pair distribution function Structure factor Significance The special values of  $\vec{q}$  where the peaks occur are those values satisfying  $\exp(i\vec{q}\cdot\vec{R}) = 1$ 

for all lattice vectors  $\vec{R}$  of the crystal lattice. These values of  $\vec{q}$  are just the reciprocal lattice vectors  $\vec{K}$ .

# Order parameter for solid

density  $\rho(r) = \sum_{i} \delta(r - r_i)$ Fourier transform  $\rho(q) = \int dr \, \rho(r) e^{iq \cdot r} / N = \sum_{j} e^{iq \cdot r_j} / N$ 

For solids , there are vectors G so that  $G \cdot R_j = 2\pi n$  for some integer n.

For example, in one dimension, for  $R_j = ja$ , G=  $2\pi/a$ 

 $\rho(G) = \sum_{j} e^{iG \cdot u_{j}} / N \text{ is a good order parameter.}$  $\rho(G) = \sum_{j} e^{iG \cdot u_{j}} / N$ 

# Average of the order parameter

- <  $\rho(G) > = \sum_j < e^{iG \cdot u_j} > /N$
- Use cumulant expansion:
- $< e^{if} > \cong e^{-\frac{\langle f^2 \rangle}{2} + \cdots}$  for any function f such that  $\langle f \rangle = 0$ .
- This can be seen by comparing the series expansion on both sides.
- $< \rho(G) > \cong \sum_{j} e^{-\langle [G \cdot u_j]^2 \rangle/2} / \mathbb{N}.$
- For fluids  $< u^2 > = \infty$ ,  $< \rho(G) > = 0$

# Fluctuation of the order parameter

• < 
$$|\rho(G)|^2$$
 > =  $\sum_{j,k}$  <  $e^{iG \cdot (u_j - u_k)}$  >

• Use cumulant expansion:

• 
$$< e^{if} > \cong e^{-\frac{\langle f^2 \rangle}{2} + \cdots}$$
 for any function f such that  $< f > = 0$ .

• This can be seen by comparing the series expansion on both sides.

• < 
$$|\rho(G)|^2$$
 >  $/N^2 \cong \sum_j e^{-\langle [G \cdot (u_0 - u_j)]^2 \rangle} / N$ 



# **Reciprocal lattice**

Scattering experiment Nature of experiment Interaction between probe and system Interaction matrix element

Fourier transform of vOne-particle density Static structure factor Typical forms of  $S(\vec{q})$ Dilute gas Liquid structure factor

Distribution function

Critical point

Crystal structure factor

#### Reciprocal lattice

Temperature variation Molecular dynamics Range of correlations Scattering divergence near critical point Compressibility rule Derivation of compressibility rule One-particle density Average values Variance of *N* Pair distribution function Structure factor Significance The special values of  $\vec{q}$  where the peaks occur are those values satisfying  $\vec{q}$ 

 $\exp(\mathrm{i}\vec{q}\cdot\vec{R})=1$ 

for all lattice vectors R<sup>-</sup> of the crystal lattice.
■ These q<sup>-</sup> are called the reciprocal lattice vectors K<sup>-</sup>. of the crystal.
■ A cumulant expansion with the lattice vibration gives the static structure has the form<sup>4</sup>

$$S(\vec{q}) = N e^{-2W(\vec{q})} \sum_{\vec{K}} \delta_{\vec{q},\vec{K}} ,$$

where W is a smoothly varying function of  $\vec{q}$  and T known as the Debye–Waller factor.

<sup>4</sup>N. W. Ashcroft and N. D. Mermin, *Solid State Physics* (Saunders, Philadelphia, 1976), pp. 790–795.



## **Temperature variation**

Scattering experiment Nature of experiment Interaction between probe and system Interaction matrix element Fourier transform of v

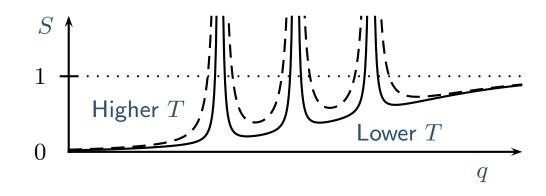
- One-particle density Static structure factor Typical forms of  $S(\vec{q})$ Dilute gas
- Liquid structure factor Distribution function
- Critical point
- Crystal structure factor

Reciprocal lattice

#### Temperature variation

Molecular dynamics Range of correlations Scattering divergence near critical point Compressibility rule Derivation of compressibility rule One-particle density Average values Variance of *N* Pair distribution function Structure factor Significance

### As the temperature increases, the peaks become broader:



This indicates a change in the motion of the system, but one that preserves the crystal symmetry.



### **Temperature variation**

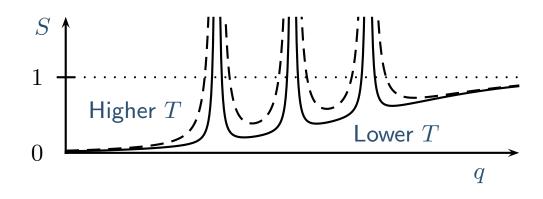
Scattering experiment Nature of experiment Interaction between probe and system Interaction matrix element

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- Distribution function
- Critical point
- Crystal structure factor Reciprocal lattice

#### Temperature variation

Molecular dynamics Range of correlations Scattering divergence near critical point Compressibility rule Derivation of compressibility rule One-particle density Average values Variance of *N* Pair distribution function Structure factor Significance

### As the temperature increases, the peaks become broader:



- This indicates a change in the motion of the system, but one that preserves the crystal symmetry.
- When T finally exceeds the melting temperature, the O(N) peaks of the crystal are reduced to the O(1) peaks of the liquid.
- However, the liquid still shows significant order even though it is a continuous phase.



# **Molecular dynamics**

Scattering experiment Nature of experiment Interaction between probe and system Interaction matrix element Fourier transform of vOne-particle density

Static structure factor Typical forms of  $S(\vec{q})$  Dilute gas

Liquid structure factor Distribution function

Critical point

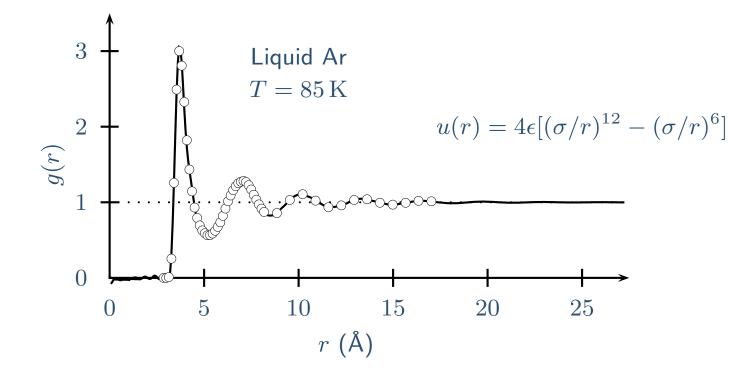
Crystal structure factor

Reciprocal lattice

Temperature variation

#### Molecular dynamics

Range of correlations Scattering divergence near critical point Compressibility rule Derivation of compressibility rule One-particle density Average values Variance of *N* Pair distribution function Structure factor Significance Molecular dynamics simulations based on a Lennard-Jones potential<sup>5</sup> are in excellent agreement with the measured g(r) of liquid argon:<sup>6</sup>



The circles in the figure are Verlet's simulation data for  $k_{\rm B}T = 0.719\epsilon$ and  $n = 0.850\sigma^{-3}$ , where  $\epsilon/k_{\rm B} = 119.8$  K and  $\sigma = 3.405$  Å.

<sup>5</sup>L. Verlet, *Phys. Rev.* **165**, 201 (1968).

<sup>6</sup>J. L. Yarnell, M. J. Katz, R. G. Wenzel, and S. H. Koenig, *Phys. Rev. A* 7, 2130 (1973).



# **Range of correlations**

Scattering experiment Nature of experiment Interaction between probe and system Interaction matrix element Fourier transform of vOne-particle density Static structure factor

Typical forms of  $S(\vec{q})$ Dilute gas Liquid structure factor Distribution function

Critical point

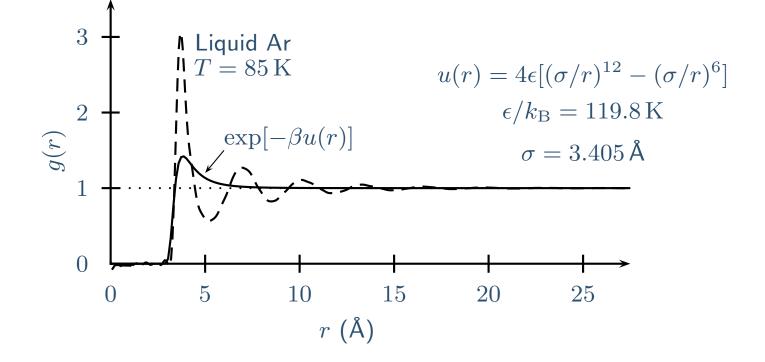
Crystal structure factor

Reciprocal lattice

Temperature variation

Molecular dynamics Range of correlations

Scattering divergence near critical point Compressibility rule Derivation of compressibility rule One-particle density Average values Variance of *N* Pair distribution function Structure factor Significance How well does the measured distribution function of a liquid agree with the approximation  $g(r) \approx \exp[-\beta u(r)]$  valid for a dilute gas?



Notice that the correlations in  $\exp[-\beta u(r)]$  die out quickly for  $r > \sigma$ , but those in the measured g(r) continue out to many  $\sigma$ .



# Scattering divergence near critical point

Scattering experiment Nature of experiment Interaction between probe and system Interaction matrix element Fourier transform of vOne-particle density Static structure factor Typical forms of  $S(\vec{q})$ Dilute gas Liquid structure factor Distribution function Critical point Crystal structure factor Pacingreal lattice

Reciprocal lattice Temperature variation Molecular dynamics Range of correlations Scattering divergence near critical point

Compressibility rule Derivation of compressibility rule One-particle density Average values Variance of *N* Pair distribution function Structure factor

Significance

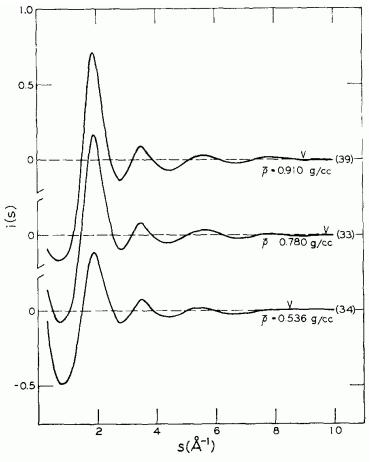


FIG. 5. The argon intensity function i(s) at  $t = -120^{\circ}$ C.

How can we understand the divergence in scattering for a fluid near the critical point?



# Scattering divergence near critical point

Scattering experiment Nature of experiment Interaction between probe and system Interaction matrix element Fourier transform of vOne-particle density Static structure factor Typical forms of  $S(\vec{q})$ Dilute gas Liquid structure factor Distribution function Critical point Crystal structure factor

Reciprocal lattice Temperature variation Molecular dynamics Range of correlations Scattering divergence near critical point

Compressibility rule Derivation of compressibility rule One-particle density Average values Variance of *N* Pair distribution function Structure factor Significance

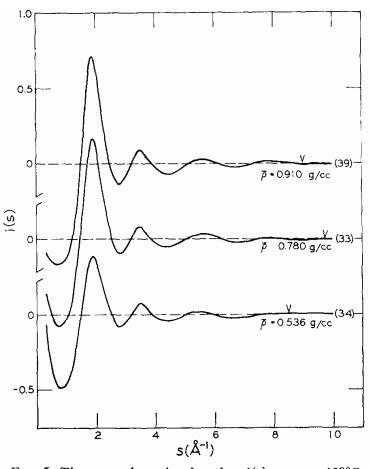


FIG. 5. The argon intensity function i(s) at t = -120°C.

- How can we understand the divergence in scattering for a fluid near the critical point?
- We have proved the following theorem for the structure factor at long wavelengths:

 $\lim_{q \to 0} S(q) = nk_{\rm B}T K_T,$ 

where  $K_T$  is the isothermal compressibility

$$K_T \equiv -\frac{1}{V} \left(\frac{\partial V}{\partial P}\right)_T.$$

This is just the inverse of the isothermal bulk modulus

$$B_T \equiv -V\left(\frac{\partial P}{\partial V}\right)_T = \frac{1}{K_T}.$$



# **Compressibility rule**

Scattering experiment Nature of experiment Interaction between probe and system Interaction matrix element Fourier transform of vOne-particle density Static structure factor Typical forms of  $S(\vec{q})$ 

Dilute gas

Liquid structure factor

Distribution function

Critical point

Crystal structure factor

Reciprocal lattice

Temperature variation

Molecular dynamics Range of correlations Scattering divergence

near critical point Compressibility rule

Derivation of compressibility rule One-particle density Average values Variance of *N* Pair distribution function Structure factor Significance The compressibility rule can also be written as

$$\lim_{q \to 0} S(q) = \frac{K_T}{K_T^{\mathrm{id}}} = \frac{B_T^{\mathrm{id}}}{B_T},$$

where  $K_T^{\text{id}}$  is the compressibility of an ideal gas at the same density n and temperature T:

$$K_T^{\rm id} = \frac{1}{nk_{\rm B}T}.$$



# **Compressibility rule**

Scattering experiment Nature of experiment Interaction between probe and system Interaction matrix element Fourier transform of vOne-particle density

Static structure factor

Typical forms of  $S(\vec{q})$ 

Dilute gas

Liquid structure factor

Distribution function

Critical point

Crystal structure factor Reciprocal lattice

Temperature variation

Molecular dynamics Range of correlations Scattering divergence

#### near critical point Compressibility rule

Derivation of compressibility rule One-particle density Average values Variance of *N* Pair distribution function Structure factor Significance

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$$\lim_{q \to 0} S(q) = \frac{K_T}{K_T^{\mathrm{id}}} = \frac{B_T^{\mathrm{id}}}{B_T},$$

where  $K_T^{\text{id}}$  is the compressibility of an ideal gas at the same density n and temperature T:

$$K_T^{\rm id} = \frac{1}{nk_{\rm B}T}.$$

In an ordinary liquid far from the critical point, the order of magnitude of these quantities is

 $B_T \sim 30000 \, \text{atm}, \quad B_T^{\text{id}} \sim 1000 \, \text{atm}; \qquad \therefore \lim_{q \to 0} S(q) \sim 0.03.$ 



# **Compressibility rule**

Scattering experiment Nature of experiment Interaction between probe and system Interaction matrix element Fourier transform of vOne-particle density

Static structure factor

Typical forms of  $S(\vec{q})$ 

Dilute gas

Liquid structure factor

Distribution function

Critical point

Crystal structure factor Reciprocal lattice

Temperature variation Molecular dynamics

Range of correlations Scattering divergence near critical point

Compressibility rule

Derivation of compressibility rule One-particle density Average values Variance of *N* Pair distribution function Structure factor Significance The compressibility rule can also be written as

$$\lim_{q \to 0} S(q) = \frac{K_T}{K_T^{\mathrm{id}}} = \frac{B_T^{\mathrm{id}}}{B_T},$$

where  $K_T^{id}$  is the compressibility of an ideal gas at the same density n and temperature T:

$$K_T^{\rm id} = \frac{1}{nk_{\rm B}T}.$$

In an ordinary liquid far from the critical point, the order of magnitude of these quantities is

 $B_T \sim 30000 \text{ atm}, \quad B_T^{\text{id}} \sim 1000 \text{ atm}; \qquad \therefore \lim_{q \to 0} S(q) \sim 0.03.$ 

However, at the critical point  $(T_c, P_c)$ , we have  $\partial P/\partial V = 0$  and thus  $B_T = 0$  (i.e.,  $K_T \to \infty$ ); consequently,

$$\lim_{q \to 0} S(q) = \infty.$$



# Significance

Scattering experiment Nature of experiment Interaction between probe and system Interaction matrix element

Fourier transform of vOne-particle density Static structure factor Typical forms of  $S(\vec{q})$ Dilute gas Liquid structure factor Distribution function Critical point Crystal structure factor **Reciprocal lattice** Temperature variation Molecular dynamics Range of correlations Scattering divergence near critical point Compressibility rule

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Significance

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$$\lim_{q \to 0} S(q) = \frac{K_T}{K_T^{\mathrm{id}}}$$

- On the left side, S(q) is a microscopic quantity describing the structural order in a fluid.
- On the right side are thermodynamic quantities.



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- However, it diverges near the critical point, where  $K_T \rightarrow \infty$ .



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- However, it diverges near the critical point, where  $K_T \rightarrow \infty$ .
- Since S(q) 1 is the Fourier transform of n[g(r) 1], a divergence of S(q) at long wavelengths

$$\lambda = \frac{2\pi}{q}$$

Pair correlation function can be measured experimentally

- Pair correlation function  $g(r, r') = \frac{n(r, r')}{n(r)n(r')}$
- In scattering experiments, one can meassure

• 
$$S(k) = \sum \exp(ik \cdot r_{jl}) / N$$

• 
$$S(k) = \int \frac{drdr'}{N} < n(r)n(r') > e^{ik(r-r')} = 1 + n \left[\frac{\delta(k)}{N} + \int (g-1)e^{ik(r-r')}\right]$$

• The compressibility  $n kT \chi = \frac{kT}{\partial p/\partial n} = 1 + n \int d^3r(g-1)$  can be realted to the zero q limit of S(q)!

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# Do you know how to compile and run a computer program

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- (2) No

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